



Research article

Synthesis, α -glucosidase inhibition, α -amylase inhibition, and molecular docking studies of 3,3-di(indolyl)indolin-2-onesMardi Santoso^{a,*}, Li Lin Ong^{b,c}, Nur Pasca Aijijiyah^a, First Ambar Wati^a, Azminah Azminah^d, Rose Malina Annur^a, Arif Fadlan^a, Zaher M.A. Judeh^{c,e}^a Department of Chemistry, Faculty of Science, Institut Teknologi Sepuluh Nopember, Sukolilo, Surabaya, 60111, Indonesia^b NTU Institute for Health Technologies, Interdisciplinary Graduate School, Nanyang Technological University, Research Techno Plaza, XFrontiers Block, #02-07, 50 Nanyang Drive, 637553, Singapore^c School of Chemical and Biomedical Engineering, Nanyang Technological University, 62 Nanyang Drive, N1.2-B1-14, 637459, Singapore^d Department of Pharmaceutical Chemistry, Faculty of Pharmacy, University of Surabaya, Surabaya, 60284, Indonesia^e Division of Chemistry & Biological Chemistry, School of Physical & Mathematical Sciences, Nanyang Technological University, 21 Nanyang Link, 637371, Singapore

ARTICLE INFO

Keywords:

3,3-Di(indolyl)indolin-2-ones

Diabetes

 α -glucosidase inhibition α -amylase inhibition

Docking studies

ABSTRACT

The synthesized 3,3-di(indolyl)indolin-2-ones **1a-p** showed desired higher α -glucosidase inhibitory activities and lower α -amylase inhibitory activities than standard drug acarbose. Particularly, compound **1i** showed favorable higher α -glucosidase % inhibition of 67 ± 13 and lower α -amylase % inhibition of 51 ± 4 in comparison to acarbose with % inhibition activities of 19 ± 5 and 90 ± 2 , respectively. Docking studies of selected 3,3-di(indolyl)indolin-2-ones revealed key interactions with the active sites of both α -glucosidase and α -amylase, further supporting the observed % inhibitory activities. Furthermore, the binding energies are consistent with the % inhibition values. The results suggest that 3,3-di(indolyl)indolin-2-ones may be developed as suitable Alpha Glucosidase Inhibitors (AGIs) and the lower α -amylase activities should be advantageous to reduce the side effects exhibited by commercial AGIs.

1. Introduction

Diabetes mellitus is a metabolic disease associated with high levels of sugars in the blood (hyperglycemia) [1]. Diabetes Type 1 accounts for ~10% and occurs in patients whose pancreas is not able to produce enough insulin. Type 2 diabetes accounts for ~90% and occurs when the body cannot use the secreted insulin effectively [2, 3]. Diabetes causes severe problems including cardiovascular diseases, nephropathy, neuropathy, retinopathy, etc [2, 3].

There are several approaches to control hyperglycemia in the blood each with its advantages and disadvantages. Such approaches mainly enhance insulin availability or remove sugar more effectively [1, 2]. An effective approach to limit the amount of sugar entering the bloodstream is to use α -glucosidase inhibitors (AGIs) [4] which inhibit the α -glucosidase enzyme. Unfortunately, commercial AGIs cause undesired strong inhibition of α -amylase enzymes and are associated with severe side effects. α -Amylase enzymes in the saliva and pancreatic juices hydrolyze α -1,4-glycosidic bonds of starch to simpler dextrans, disaccharides, and oligosaccharides [5, 6]. As the food travels to the

small intestine, the brush border cells of the epithelium secrete α -glucosidase enzymes which hydrolyze disaccharides and oligosaccharides giving α -D-glucose that pass to the bloodstream [7]. Consequently, inhibition of α -glucosidase limits the amount of α -D-glucose that passes to the bloodstream [8]. Current commercial AGI drugs represented by Acarbose, Miglitol, and Voglibose are effective inhibitors of α -glucosidase. However, they cause flatulence, diarrhea, bloating, abdominal pain, and discomfort [4]. It is thought that the side-effects result from fermentation of undigested carbohydrates due to strong inhibition of α -amylase [9]. Therefore, it is desirable to design new selective AGIs having strong inhibition of α -glucosidase and low inhibition of α -amylase.

Indole-based compounds are abundant in the plant kingdom and show many bioactivities including antimalarial [10], antifungal [11], anticancer [12], antibacterial [12], anti-diabetic [13], antihelmintic [14] activities. Several indole-based drugs such as Sunitinib, indolidan, delaverdine, indomethacin, indoxole, and vinblastine are marketed for the treatment of various diseases while others are at various stages of clinical trials [15, 16, 17].

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Received 6 June 2021; Received in revised form 3 August 2021; Accepted 1 March 2022

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Oxindoles also exhibit many activities including antiviral, antimicrobial, antifungal, anticancer, anti-inflammatory, antihypertensive, serotonergic, analgesic and sleep-inducing activities [17]. Oxindoles of the 3,3-di(indolyl)indolin-2-ones type (Scheme 1) show antidiabetic [18], antimicrobial [19], anticancer [20], and spermicidal [21] activities. Recently, Wang and co-workers [18] reported promising α -glucosidase inhibition activities of several 3,3-di(indolyl)indolin-2-ones. However, the authors did not examine the α -amylase inhibition activities which are important for developing next-generation AGIs having minimal gastrointestinal side-effects. They also did not elucidate the structure-activity relationship of 3,3-di(indolyl)indolin-2-ones with aromatic substituents as well as strongly donating and withdrawing moieties on inhibition of α -glucosidase. Therefore, at this stage, the overall inhibitory activity profile of 3,3-di(indolyl)indolin-2-ones remains unclear.

Herein, we report the synthesis of diverse 3,3-di(indolyl)indolin-2-ones and examine their α -glucosidase and α -amylase activities to provide a clear understanding of their overall inhibition effectiveness. To complement our study, we also report molecular docking studies to elucidate the mechanism of action of these compounds.

2. Materials and methods

The starting materials, solvents, and reagents were purchased from Sigma-Aldrich, Merck, and Fluka, and were used without further purification unless stated. Thin-layer chromatography was performed on Merck 0.20 mm precoated silica gel aluminum plates (Kieselgel 60, F254) and visualized using a UV lamp operating at 245 nm. Melting points were measured using the Fischer-John apparatus and are uncorrected. Infrared spectra were recorded using KBr disc on FTIR Shimadzu 8400S. ^1H NMR spectra were recorded using Jeol JNM-ECA300 (300 MHz), Jeol JNM-ECS400 (400 MHz), Bruker Avance DPX 300 (300 MHz), or Hitachi R-1900 FT NMR (90 MHz). ^{13}C NMR spectra were recorded at 75 MHz on Jeol JNM-ECA300 (300 MHz), 100 MHz on Jeol JNM-ECS400 (400 MHz), and 75.47 MHz on a Bruker Avance DPX 300. Mass spectra were obtained using a Xevo G2-XS QToF, Hitachi QP-5000, or Waters LCT Premier XE instrument.

2.1. Chemistry

2.1.1. General procedure for the synthesis of 3,3-di(indolyl)indolin-2-one

A solution of the isatin or its alkyl derivative and indole or its alkyl derivative in methanol (~100 ml per 1g of isatin) was treated with a

catalytic amount of BF_3 or H_2SO_4 (2–3 drops per 1 g of isatin) and stirred at 40–60 °C for 1–2 h. After completion of the reaction as indicated by TLC, the reaction mixture was diluted with ice-cold water (50 mL per 1g of isatin). The resulting precipitates were filtered under vacuum, washed with an excess of ice-cold water (3 x 50 mL), and then were dried under vacuum to give the pure product. This general procedure was used to prepare 3,3-di(indolyl)indolin-2-one derivatives (**1a–1c**, **1f**, **1i–1p**).

3,3-Di(1H-indole-3-yl)indolin-2-one (1a): Isatin (0.30 g, 2.04 mmol) and indole (0.47 g, 4.00 mmol) reacted to give **1a** as white powder, 0.71 g, 97% yield, m.p. 320–321 °C (lit [22]. 317–319 °C). ^1H NMR (90 MHz, $\text{DMSO}-d_6$): δ 6.70–7.40 (m, 14H), 10.54 (s, 2H), 10.91 (s, 1H). Mass spectrum (ED): m/z 363 (M, 80%), 334 (100), 247 (10), 219 (50).

3,3-Di(1-methyl-1H-indole-3-yl)indolin-2-one (1b): Isatin (0.25 g, 1.70 mmol) and 1-methylindole (0.44 g, 3.35 mmol) reacted to give **1b** as white powder (0.58 g, 88% yield), m.p. 329–330 °C (lit [23]. 330–332 °C). ^1H NMR (90 MHz, $(\text{CD}_3)_2\text{CO}$): δ 3.76 (s, 6H), 6.74–7.45 (m, 14H), 9.45 (bs, 1H). Mass spectrum (ED): m/z 392 (M+1, 10%), 391 (M, 60), 376 (5), 362 (100), 233 (20).

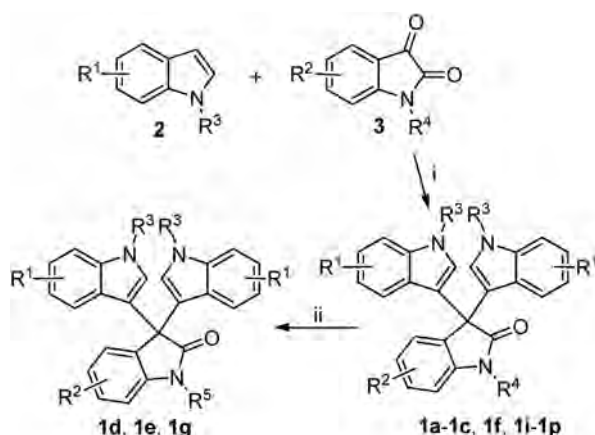
3,3-Di(1-ethyl-1H-indole-3-yl)indolin-2-one (1c): Isatin (0.052 g, 0.35 mmol) and 1-ethylindole (0.10 g, 0.69 mmol) reacted to give **1c** as white powder (0.13 g, 93% yield), m.p. 340–341 °C (lit [24]. mp 273–275 °C). ^1H NMR (90 MHz, $(\text{CD}_3)_2\text{CO}$): δ 1.28 (t, $J = 7.2$ Hz, 6H), 4.14 (q, $J = 7.2$ Hz, 4H), 6.73–7.47 (m, 14H), 10.56 (s, 1H). Mass spectrum (ED): m/z 420 (M+1, 10%), 419 (M, 65), 390 (100), 275 (15), 247 (10).

3,3-Di(1H-indole-3-yl)-5-bromoindolin-2-one (1f): 5-Bromoisatin (0.050 g, 0.22 mmol) and indole (0.050 g, 0.45 mmol) reacted to give **1f** as brown solid (0.090 g, 92% yield), m.p. 282–283 °C (lit [19]. 299–301 °C). ^1H NMR (400 MHz, $(\text{CD}_3)_2\text{SO}$): δ 6.80 (t, $J = 7.3$ Hz, 2H), 6.87 (s, 2H), 6.95 (d, $J = 8.2$ Hz, 1H), 7.01 (t, $J = 7.2$ Hz, 2H), 7.18 (d, $J = 7.8$ Hz, 2H), 7.28 (s, 1H), 7.35 (d, $J = 8.0$ Hz, 2H), 7.39 (d, $J = 8.3$ Hz, 1H), 10.74 (s, 1H), 10.99 (s, 2H). Mass spectrum (EI): m/z 443 (M, ^{81}Br , 100%), 441 (M, ^{79}Br , 10), 327 (5), 325 (5), 299 (5), 297 (5), 44 (100).

3,3-Di(5-hydroxy-1H-indole-3-yl)-5-nitroindolin-2-one (1i): 5-Nitroisatin (0.096 g, 0.50 mmol) and 5-hydroxyindole (0.13 g, 1.00 mmol) reacted, and after column chromatography (ethyl acetate:*n*-hexane, 2:1), gave **1i** as brown solid (0.15 g, 68 % yield), m.p. 302–303 °C. IR (KBr disk): 3524, 3349, 3117, 1697, 1626, 1605, 1582, 1528, 1468, 1338, 1202, 1062 cm^{-1} . ^1H NMR (400 MHz, $\text{DMSO}-d_6$): δ 6.53 (dd, $J = 8.4$, $J = 2.4$ Hz, 4H), 6.57 (d, $J = 2.4$ Hz, 2H), 6.78 (d, $J = 2.4$ Hz, 2H), 7.12–7.16 (m, 3H), 7.90 (d, $J = 2.0$ Hz, 1H), 8.20 (dd, $J = 8.4$, $J = 2.4$ Hz, 1H), 8.52 (s, 2H), 10.72 (s, 2H), 11.20 (bs, 1H). ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$): δ 53.1, 105.1, 110.3, 112.2, 112.6, 120.5, 125.6, 125.7, 125.9, 126.7, 132.1, 136.0, 142.7, 148.4, 150.5, 179.5.

3,3-Di(1H-indole-3-yl)-5-nitroindolin-2-one (1j): 5-Nitroisatin (0.072 g, 0.38 mmol) and indole (0.088 g, 0.75 mmol) reacted to give **1j** as yellow solid (0.13 g, 87% yield), m.p. 272–273 °C (lit [22]. 281–282 °C). IR (KBr disc): 3385, 3123, 3059, 1709, 1620, 1528, 1481, 1456, 1420, 1340, 1246, 1215, 1175, 1129, 1086, 1013, 941 cm^{-1} . ^1H NMR (300 MHz, $(\text{CD}_3)_2\text{CO}$): δ 6.87 (t, $J = 4.6$ Hz, 2H), 7.05–7.09 (m, 4H), 7.07 (s, 1H), 7.32 (d, $J = 9.1$ Hz, 1H), 7.42 (d, $J = 5.1$ Hz, 4H), 7.44 (d, $J = 4.8$ Hz, 2H), 8.15 (d, $J = 1.5$ Hz, 1H), 8.27 (dd, $J = 4.0$, 1.5 Hz, 1H), 10.20 (bs, 1H), 10.29 (s, 2H). ^{13}C NMR (75 MHz, $(\text{CD}_3)_2\text{CO}$): δ 53.9, 110.6, 112.6, 114.7, 119.8, 121.6, 121.9, 122.5, 125.6, 126.1, 126.9, 136.6, 138.5, 143.9, 148.6, 179.5. MS (ES): m/z calcd for $\text{C}_{24}\text{H}_{15}\text{N}_4\text{O}_3$, [M-H] $^-$ 407.1144; Found 407.1124.

3,3-Di(1-methyl-1H-indole-3-yl)-5-nitroindolin-2-one (1k): 5-Nitroisatin (0.29 g, 1.51 mmol) and 1-methylindole (0.39 g, 2.97 mmol) reacted to give **1k** as yellow solid (0.61 g, 94% yield), m.p. 291–292 °C (lit [25]. >300 °C). ^1H NMR (300 MHz, $(\text{CD}_3)_2\text{CO}$): δ 3.78 (s, 6H), 6.88 (t, $J = 4.1$ Hz, 2H), 7.00 (s, 2H), 7.12 (t, $J = 4.6$ Hz, 2H), 7.31 (d, $J = 5.2$ Hz, 1H), 7.37 (d, $J = 4.8$ Hz, 2H), 7.41 (d, $J = 4.8$ Hz, 2H), 8.12 (d, $J = 1.5$ Hz, 1H), 8.26 (dd, $J = 4.5$, 1.4 Hz, 1H), 10.29 (bs, 2H). ^{13}C NMR (125 MHz, $(\text{CD}_3)_2\text{CO}$): δ 32.9, 53.7, 110.6, 110.6, 113.6, 119.7, 121.6, 122.1, 122.5, 126.1, 127.3, 129.8, 136.6, 138.9, 143.9, 148.6, 179.4. MS (ES): m/z calcd for $\text{C}_{26}\text{H}_{19}\text{N}_4\text{O}_3$, [M-H] $^-$ 435.1457; Found 435.1436.



Reagents and conditions: (i) BF_3 or H_2SO_4 , MeOH, 2 h, 40–60 °C; (ii) MeI, or EtI or Me_2SO_4 , KOH, DMSO , 2 h, 0 °C to r.t.

Scheme 1. Synthesis of 3,3-di(indolyl)indolin-2-ones **1a–1p**.

3,3-Di(1*H*-indole-3-yl)-1-benzyl-5-nitroindolin-2-one (**1l**): 1-Benzyl-5-nitroisatin (0.070 g, 0.25 mmol) and indole (0.060 g, 0.51 mmol) reacted to give **1l** as a yellow solid (0.10 g, 83% yield), m.p. 237–238 °C. IR (KBr disc): 3458, 3352, 3125, 3061, 1711, 1603, 1518, 1487, 1454, 1335, 1211, 1173, 1105, 1080, 748 cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): δ 5.09 (s, 2H), 6.74 (t, *J* = 7.6 Hz, 2H), 6.95 (d, *J* = 2.4 Hz, 2H), 7.01 (d, *J* = 7.6 Hz, 2H), 7.07 (d, *J* = 8.6 Hz, 2H), 7.28–7.37 (m, 8H), 8.00 (d, *J* = 2.4 Hz, 1H), 8.24 (dd, *J* = 8.6, 2.4 Hz, 1H), 11.09 (s, 2H). ¹³C NMR (400 MHz, DMSO-*d*₆): δ 43.9, 52.7, 110.2, 112.4, 113.0, 119.1, 120.4, 120.9, 121.8, 125.2, 125.8, 125.9, 128.2, 128.3, 129.3, 135.0, 136.3, 137.6, 143.4, 148.2, 177.9. HRMS (ES): *m/z* calcd for C₃₁H₂₁N₄O₃, [M-H]⁻ 497.1614; Found 497.1636.

3,3-Di(1*H*-indole-3-yl)-1-benzyl-5-bromoindolin-2-one (**1m**): 1-Benzyl-5-bromoisatin (0.030 g, 0.10 mmol) and indole (0.020 g, 0.20 mmol) reacted to give **1m** as yellow solid (0.043 g, 81% yield), m.p. 188–189 °C (lit [19]). 268–270 °C. IR (KBr disc): 3428, 3356, 3131, 3063, 1699, 1603, 1546, 1479, 1454, 1422, 1341, 1242, 1213, 1173, 1101, 1065, 1013, 749 cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): δ 5.00 (s, 2H), 6.74 (t, *J* = 8.0 Hz, 2H), 6.89 (d, *J* = 2.8 Hz, 2H), 7.01 (t, *J* = 7.6 Hz, 2H), 7.08 (t, *J* = 7.0 Hz, 3H), 7.26–7.45 (m, 10H), 11.03 (s, 2H). ¹³C NMR (400 MHz, DMSO-*d*₆): δ 43.6, 53.0, 112.0, 112.3, 113.7, 114.7, 119.0, 121.0, 121.7, 125.0, 125.9, 127.8, 128.1, 129.2, 131.3, 136.7, 136.7, 137.5, 141.6, 177.1. MS (ES): *m/z* calcd for C₃₁H₃₁N₃OBr, [M-H]⁻ 530.0868; Found 530.0910.

3,3-Di(1*H*-indole-3-yl)-1-(4-bromobenzyl)-5-bromoindolin-2-one (**1n**): 1-(4-Bromobenzyl)-5-bromoisatin (0.19 g, 0.48 mmol) and indole (0.12 g, 1.02 mmol) reacted to give **1n** as ivory coloured solid (0.28 g, 97% yield), m.p. 248–249 °C. IR (KBr disc): 3416, 3287, 3056, 1715, 1601, 1537, 1422, 1333, 1244, 1206, 1167, 1101, 1071, 1013, 741 cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): δ 4.97 (s, 2H), 6.73 (t, *J* = 7.4 Hz, 2H), 6.87 (s, 2H), 7.00 (dd, *J* = 12.8, 8.0 Hz, 4H), 7.13 (d, *J* = 8.0 Hz, 1H), 7.28 (d, *J* = 8.0 Hz, 2H), 7.35 (d, *J* = 7.4 Hz, 3H), 7.45 (d, *J* = 8.4 Hz, 1H), 7.49 (d, *J* = 8.0 Hz, 2H), 11.02 (s, 2H). ¹³C NMR (400 MHz, DMSO-*d*₆): δ 42.5, 52.4, 111.5, 111.9, 113.1, 114.3, 118.5, 120.5, 120.8, 121.2, 124.6, 125.4, 127.3, 130.0, 130.8, 131.6, 135.7, 136.1, 137.0, 140.9, 176.6. MS (ES): *m/z* calcd for C₃₁H₂₁N₃OBr₂, [M + Na]⁺ 633.9929; Found 633.9542.

3,3-Di(1*H*-indole-3-yl)-5-chloroindolin-2-one (**1o**): 5-Chloroisatin (0.051 g, 0.28 mmol) and indole (0.065 g, 0.56 mmol) reacted to give **1o** as brown solid (0.10 g, 98% yield), m.p. 275–276 °C (lit [22]). 290–291 °C. ¹H NMR (90 MHz, CDCl₃): δ 6.81–7.42 (m, 13H), 10.70 (s, 1H), 10.98 (s, 2H). Mass spectrum (EI): *m/z* 399 (M, ³⁷Cl, 10%), 397 (M, ³⁵Cl, 30), 368 (15), 246 (50), 117 (35), 44 (100).

3,3'-Di(1-methyl-1*H*-indole-3-yl)-5-chloroindolin-2-one (**1p**): 5-Chloroisatin (0.083 g, 0.46 mmol) and 1-methylindole (0.12 g, 0.92 mmol) reacted to give **1p** as pink solid (0.17 g, 89% yield), m.p. 278–279 °C (lit [26]). >300 °C. IR (KBr disc): 3223, 2932, 1717, 1616, 1541, 1476, 1429, 1371, 1333, 1290, 1248, 1200, 739 cm⁻¹. ¹H NMR (90 MHz, DMSO-*d*₆): δ 3.72 (s, 6H), 6.77–7.44 (m, 3H), 10.74 (s, 1H). Mass spectrum (EI): *m/z* 427 (M, ³⁷Cl, 5), 425 (M, ³⁵Cl, 15), 399 (30), 397 (10), 269 (9), 267 (3), 44 (100).

2.1.2. General procedures for the synthesis of substituted 3,3-di(indolyl)indolin-2-one through *N*-alkylation

A mixture of the 3,3-di(indolyl)indolin-2-one and freshly crushed KOH in anhydrous DMSO was stirred at room temperature for 1 h. After cooling to ice-bath temperature, the alkylating agent was added and the mixture was stirred further at room temperature for 1 h. After completion of the reaction as indicated by TLC, the reaction mixture was diluted with ice-cold water (~50 mL per 0.5 mmol of starting indolin-2-one). The resulting precipitate was filtered under vacuum, washed with an excess of ice-cold water (3 x 50 mL), and dried to give the pure product. This general procedure was used to make compounds **1d**, **1e**, and **1g**.

3,3-Di(1-methyl-1*H*-indole-3-yl)-1-methylindolin-2-one (**1d**): A mixture of **1b** (0.20 g, 0.51 mmol) and KOH (0.11 g, 1.96 mmol) in anhydrous DMSO (15 mL) reacted with methyl iodide (0.06 mL, 0.96

mmol) to give **1d** as white solid (0.19 g, 90% yield), m.p. 218–219 °C (lit [23]). 232–234 °C. ¹H NMR (CDCl₃): δ 3.32 (s, 3H), 3.66 (s, 6H); 6.82–7.48 (m, 14H). Mass spectrum (EI): *m/z* 406 (M+1, 30%), 405 (M, 100), 390 (10), 376 (90), 275 (40), 247 (70), 233 (20).

3,3-Di(1-ethyl-1*H*-indole-3-yl)-1-ethylindolin-2-one (**1e**): A mixture of **1c** (0.10 g, 0.24 mmol) and KOH (0.066 g, 1.18 mmol) in anhydrous DMSO (10 mL) reacted with ethyl iodide (0.04 mL, 0.48 mmol) to give **1e** as white solid (0.10 g, 91% yield), m.p. 169–170 °C (lit [27]). 272–274 °C. ¹H NMR (CDCl₃): δ 1.23–1.29 (m, 9H), 3.83 (q, *J* = 6.0 Hz, 2H), 4.13 (q, *J* = 6.0 Hz, 4H), 6.83 (t, *J* = 9.0 Hz, 2H), 6.92 (s, 2H), 6.98–7.10 (m, 3H), 7.17–7.27 (m, 3H), 7.32 (d, *J* = 6.0 Hz, 2H), 7.43 (d, *J* = 9.0 Hz, 2H). Mass spectrum (EI): *m/z* 448 (M+1, 30%), 447 (M, 100), 418 (75), 404 (30), 390 (20), 303 (40), 275 (40), 247 (70), 233 (20).

3,3-Di(5-bromo-1-methyl-1*H*-indole-3-yl)-5-bromo-1-methylindolin-2-one (**1g**): A mixture of 3,3-di(5-bromo-1-methyl-1*H*-indole-3-yl)-5-bromoindolin-2-one (0.030 g, 0.048 mmol) and KOH (0.036 g, 0.64 mmol) in anhydrous DMSO (15 mL) reacted with dimethyl sulfate (0.06 mL, 0.63 mmol) to give **1g** as light orange solid (0.029 g, 94% yield), m.p. 287–288 °C. IR (KBr disc): 2918, 1713, 1607, 1537, 1474, 1422, 1368, 1337, 1273, 1219, 1144, 1094, 1049, 789 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): δ 3.25 (s, 3H), 3.73 (s, 6H), 7.02 (s, 2H), 7.21 (d, *J* = 8.4 Hz, 1H), 7.23 (d, *J* = 2.0 Hz, 1H), 7.25 (d, *J* = 2.0 Hz, 1H), 7.27 (d, *J* = 2.0 Hz, 1H), 7.31 (d, *J* = 2.0 Hz, 1H), 7.43 (d, *J* = 9.1 Hz, 2H), 7.59 (dd, *J* = 8.4, 2.0 Hz, 1H). ¹³C NMR (125 MHz, CDCl₃): δ 26.5, 32.7, 51.7, 111.2, 111.6, 111.7, 112.4, 114.4, 122.5, 123.9, 127.2, 130.2, 131.2, 134.8, 136.2, 141.8, 175.9. Mass spectrum (ES): *m/z* calcd for C₂₇H₂₁Br₃N₃O, [M + H]⁺ 639.9235; Found 639.9280.

2.1.3. Reduction of **1j** to **1h**

3,3'-Di(1*H*-indole-3-yl)-5-5-aminoindolin-2-one (**1h**): A mixture of **1j** (0.10 g, 0.25 mmol) and Pd/C (0.01 g) in ethanol (10 mL) was brought to reflux. To this mixture, a solution of hydrazine hydrate (0.49 mL, 0.01 mmol) in ethanol (10 mL) was added dropwise. The resulting mixture was heated to reflux for 90 min, allowed to cool and then diluted with THF (5 mL). The cooled mixture was filtered and the mother liquor was diluted with water (50 mL). The resulting precipitates were filtered under vacuum, washed with excess of ice-cold water (3 x 50 mL), dried over MgSO₄, purified by column chromatography (chloroform:ethyl acetate, 1:2) to give **1h** as brown-yellow solid (0.068 g, 72% yield), m.p. 258–259 °C. IR (KBr disc): 3626, 3364, 3315, 3121, 3057, 1666, 1614, 1491, 1456, 1421, 1335, 1244, 1207, 1103, 1014 cm⁻¹. ¹H NMR (400 MHz, DMSO-*d*₆): δ 5.11 (bs 2H), 6.44 (dd, *J* = 8.0, 2.0 Hz, 1H), 6.52 (s, 1H), 6.66 (d, *J* = 8.8 Hz, 1H), 6.75 (t, *J* = 7.6 Hz, 2H), 6.79 (d, *J* = 2.4 Hz, 2H), 6.97 (t, *J* = 7.2 Hz, 2H), 7.20 (d, *J* = 8.0 Hz, 2H), 7.30 (d, *J* = 7.6 Hz, 2H), 10.15 (s, 1H), 10.88 (s, 2H). ¹³C NMR (100 MHz, DMSO-*d*₆): δ 53.4, 110.3, 112.0, 113.1, 114.0, 115.3, 118.6, 121.4, 121.5, 124.8, 126.3, 132.3, 136.0, 137.4, 142.8, 179.0. Mass spectrum (ES): *m/z* calcd for C₂₄H₁₉N₄O, [M + H]⁺ 379.1559; Found 379.3989.

2.2. α-Glucosidase inhibition assay

In principle, Kang and co-workers method was used to examine the α-glucosidase inhibitory activity of the synthesized compounds [28]. In this method, a solution of 50 μg/mL final concentration was prepared for each compound in DMSO. 8 μL of this solution was added to 115 μL of 0.1 M sodium phosphate buffer pH 7.0 in a 96 well microtiter plate. To this mixture, 50 μL of α-glucosidase enzyme solution (0.5 U/mL of yeast α-glucosidase in buffer solution) was introduced. The 'Blank', was prepared similarly but 50 μL of buffer was added instead of the enzyme, and 8 μL of DMSO was added instead of the inhibitor. Positive control was also prepared in a similar fashion where no inhibitor is replaced by 8 μL of DMSO. Acarbose was used as the standard drug in this procedure. The microtiter plate was shaken for 15 min at 37 °C then 25 μL of 2.5 mM of 4-nitrophenyl β-D-glucopyranoside (PNPG) was introduced to each well. The plate was shaken for another 15 min at the same temperature and the

absorbance was measured at 405 nm. The percentage of inhibition was calculated as Eq. (1):

$$\left(\frac{Abs_{positive\ control} - Abs_{compound}}{Abs_{positive\ control}} \right) \times 100\% \quad (1)$$

All measurements are performed in triplicates and the values are represented as mean \pm standard deviation.

2.3. α -Amylase inhibition assay

Phan and co-workers method was used for testing α -amylase inhibitory activity [29]. Solution corresponding to 50 μ g/mL in DMSO was prepared. To each test tube, 50 μ L of these test solutions, 100 μ L of α -amylase (5U/mL of porcine pancreatic α -amylase in 0.05 sodium phosphate buffer pH 6.8) and 460 μ L of 0.05 M sodium phosphate buffer pH 6.8 were added. 'Blank' and 'Positive Control' were prepared similarly but 50 μ L of DMSO was added instead of the inhibitor. In the case of 'Blank', the enzyme solution was replaced by 100 μ L of the buffer. Results were compared to Acarbose as the standard drug. The test tubes were shaken for 10 min at 37 $^{\circ}$ C whereupon 450 μ L of 0.5% starch solution was added to each tube and the shaking continued for another 20 min. At this point, 500 μ L of dinitro salicylic acid (DNSA) reagent was added to each tube. The tubes were introduced in a boiling water bath for 15 min and the absorbance was then measured at 540 nm. The 0.5% starch solution was prepared by dissolving 0.5 g of starch in 99.5 mL of buffer. The starch solution was heated in a boiling water bath for 5 min to ensure starch dissolved completely. The percentage inhibition was calculated using Eq. (1). All measurements were performed in triplicates and the values are represented as mean \pm standard deviation. Note that the DNSA solution is prepared by dissolving 1 g of DNSA in 50.0 mL of ultrapure water followed by slow addition of 30.0 g of sodium tartrate and 20.0 mL of 2N NaOH solution. Finally, the volume was topped up to 100 mL with ultrapure water.

2.4. Molecular docking

Docking simulations were performed to reveal the binding modes of 3,3-di(indolyl)indolin-2-ones **1a**, **1i**, and **1n** with α -glucosidase and α -amylase enzymes. The structure of the human lysosomal α -glucosidase [30] and human pancreatic α -amylase [31] was directly

downloaded from the protein data bank (PDBID: 5NN5 and 6OCN, respectively) and was optimized after removing the co-factors, water molecules, and heteroatoms. The structures of the ligands were directly drawn using MarvinSketch program [32]. The docking simulation was conducted using Autodock 4.2.6 software [33]. The docking center was set to be the center of protein and the docking pocket size was set to be large enough to cover the whole protein molecule. The docking results were visualized using Discovery studio (Dassault Systèmes, San Diego) [34].

3. Results and discussions

3.1. Synthesis of 3,3-di(indolyl)indolin-2-ones **1a-p**

The route to synthesize 3,3-di(indolyl)indolin-2-ones **1a-p** is shown in Scheme 1. The reaction between indoles **2** and isatins **3** in the presence of a catalytic amount of BF_3 or H_2SO_4 at 40–60 $^{\circ}$ C afforded the desired products **1a-1c**, **1f**, and **1i-1p** in high 68–97% yields within 2 h. Compounds **1d**, **1e**, and **1g** were obtained conveniently in high 90–94% yields by alkylating their corresponding counterparts. The 3,3-di(indolyl)indolin-2-ones **1a-p** were characterized using MS, IR, 1H NMR and ^{13}C NMR and the spectral data were consistent with the structures and are typical for oxindole systems.¹⁸ For an illustrative example, the 1H NMR spectrum of **1n** recorded in DMSO-*d*₆ showed 21 protons. Singlet signals at δ_H 4.97 and 11.02 ppm were assigned to methylene group protons and –NH protons respectively. The ^{13}C NMR spectrum showed signals at δ_C 42.5 and 52.4 ppm correspond to a methylene carbon and a spiro carbon respectively, and a signal at δ_C 176.6 ppm indicates the presence of carbonyl carbon. The IR spectrum of compound **1n** showed peak at 1715 revealing the presence of lactam amide (–NH–CO–). MS data was acquired in the positive ionization mode, exhibited peak at $m/z = 633.9542$ $[M + Na]^+$.

3.2. α -Glucosidase and α -amylase inhibition studies

The α -glucosidase and α -amylase % inhibition activities of the synthesized 3,3-di(indolyl)indolin-2-ones **1a-p** along with that of acarbose as the positive standard drug are summarized in Table 1. In general, while the compounds showed high to excellent inhibition activities for both enzymes, they showed stronger α -glucosidase inhibitory activity and desired lower α -amylase inhibitory activity in comparison to standard AGI drug acarbose.

Table 1. Percentage inhibition of α -glucosidase and α -amylase by 3,3-di(indolyl)indolin-2-ones **1a-p** with acarbose as the reference standard.

No	Indolin-2-one	R ¹	R ²	R ³	R ⁴ /R ⁵	% α -glucosidase inhibition ^a	% α -amylase inhibition ^a
1	1a	H	H	H	H	16 \pm 6	92 \pm 4
2	1b	H	H	CH ₃	H	37 \pm 11	81 \pm 6
3	1c	H	H	CH ₃ CH ₂	H	73 \pm 6	72 \pm 5
4	1d	H	H	CH ₃	CH ₃	86 \pm 7	77 \pm 8
5	1e	H	H	CH ₃ CH ₂	CH ₃ CH ₂	50 \pm 11	74 \pm 9
6	1f	H	Br	H	H	76 \pm 8	86 \pm 10
7	1g	Br	Br	CH ₃	CH ₃	61 \pm 1	79 \pm 4
8	1h	H	NH ₂	H	H	17 \pm 3	79 \pm 9
9	1i	OH	NO ₂	H	H	67 \pm 13	51 \pm 4
10	1j	H	NO ₂	H	H	79 \pm 5	93 \pm 6
11	1k	H	NO ₂	CH ₃	H	86 \pm 6	91 \pm 5
12	1l	H	NO ₂	H	Benzyl	76 \pm 8	86 \pm 0
13	1m	H	Br	H	Benzyl	92 \pm 3	80 \pm 3
14	1n	H	Br	H	4-Br-benzyl	94 \pm 3	73 \pm 5
15	1o	H	Cl	H	H	83 \pm 2	87 \pm 6
16	1p	H	Cl	CH ₃	H	84 \pm 2	81 \pm 7
17	Acarbose ^a					19 \pm 5	90 \pm 2

^a Inhibition was measured at a concentration of 50 μ g/ml. Inhibition values are expressed as means \pm SD; n = 3.

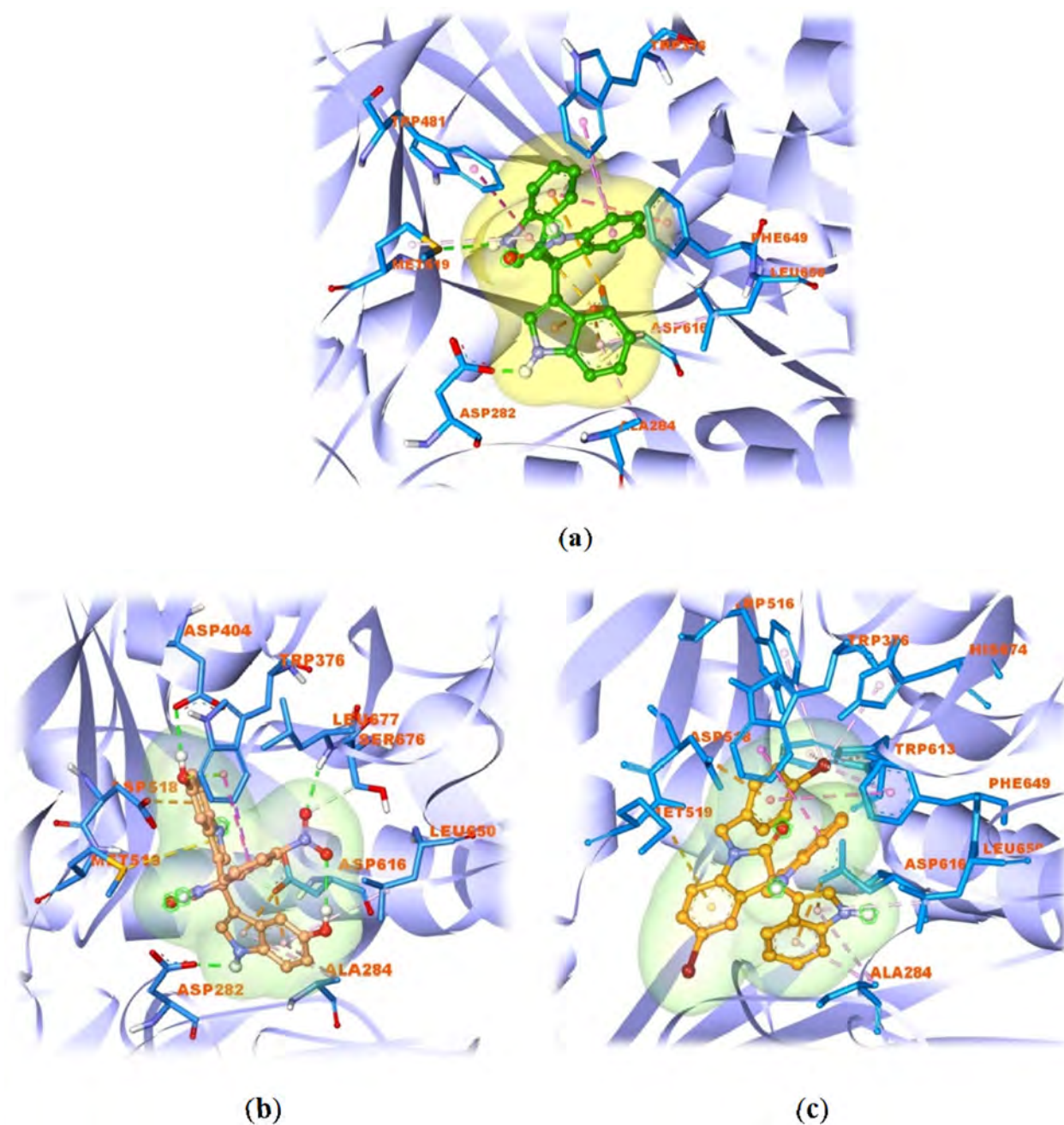


Figure 1. Binding interaction of (a) indolin-2-one **1a**; (b) indolin-2-one **1i**; and (c) indolin-2-one **1n** in with α -glucosidase (PDB ID: 5NN5).

3.2.1. α -Glucosidase inhibition activities

All the compounds showed higher % α -glucosidase inhibition activities ranging from 37 ± 11 to 94 ± 3 in comparison to acarbose with % inhibition activity of 19 ± 5 , all measured at a concentration of $50 \mu\text{g/ml}$. Exceptions are compounds **1a** and **1h** which showed % inhibition activities of 16 ± 6 and 17 ± 3 , respectively (Table 1, entries 1 and 8). Considering the compounds in Table 1, the type of substituent and their position played a role in the inhibition activities to different extents. Compound **1a** with no substituents showed the lowest % inhibition activity of 16 ± 6 . In the study conducted by Wang *et al.* [18], the same compound **1a** showed the lowest IC_{50} value of $145.95 \pm 0.46 \mu\text{M}$. Introduction of a methyl moiety at the indole rings as in **1b** lead to doubling of the % inhibition activity to 37 ± 11 while the introduction of an ethyl moiety as in **1c** increases the activity to more than four folds to 73 ± 6 . Interestingly, while the introduction of methyl moiety to the

oxindole ring increased the inhibition activity of **1d** to 86 ± 7 , the introduction of a corresponding ethyl moiety as in **1e** reduced the activity to 50 ± 11 . This unpredicted result underscores the effect of small structural changes on inhibitory activity. Compound **1f** with bromine moiety enhanced the % inhibition activity of its parent **1a** by five-folds from 16 ± 6 to 76 ± 8 . However, compound **1g** with the bromine at the same position but with another at the indole rings gave lower inhibition activity of 61 ± 1 in comparison to its parent **1d** with inhibition activity of 86 ± 7 . The introduction of strong electron-donating NH_2 group on the oxindole ring of **1a** to give **1h** did not significantly affect the % inhibition activity (17 ± 3 vs 16 ± 6 , entry 8 vs entry 1, Table 1). However, the introduction of additional OH groups on the indole rings of **1h** and replacing its NH_2 group with NO_2 group to give **1i**, significantly increased the % inhibition activity from 17 ± 3 to 67 ± 13 (Table 1, entry 8 vs entry 9). Additionally, stronger electron-withdrawing NO_2 groups of

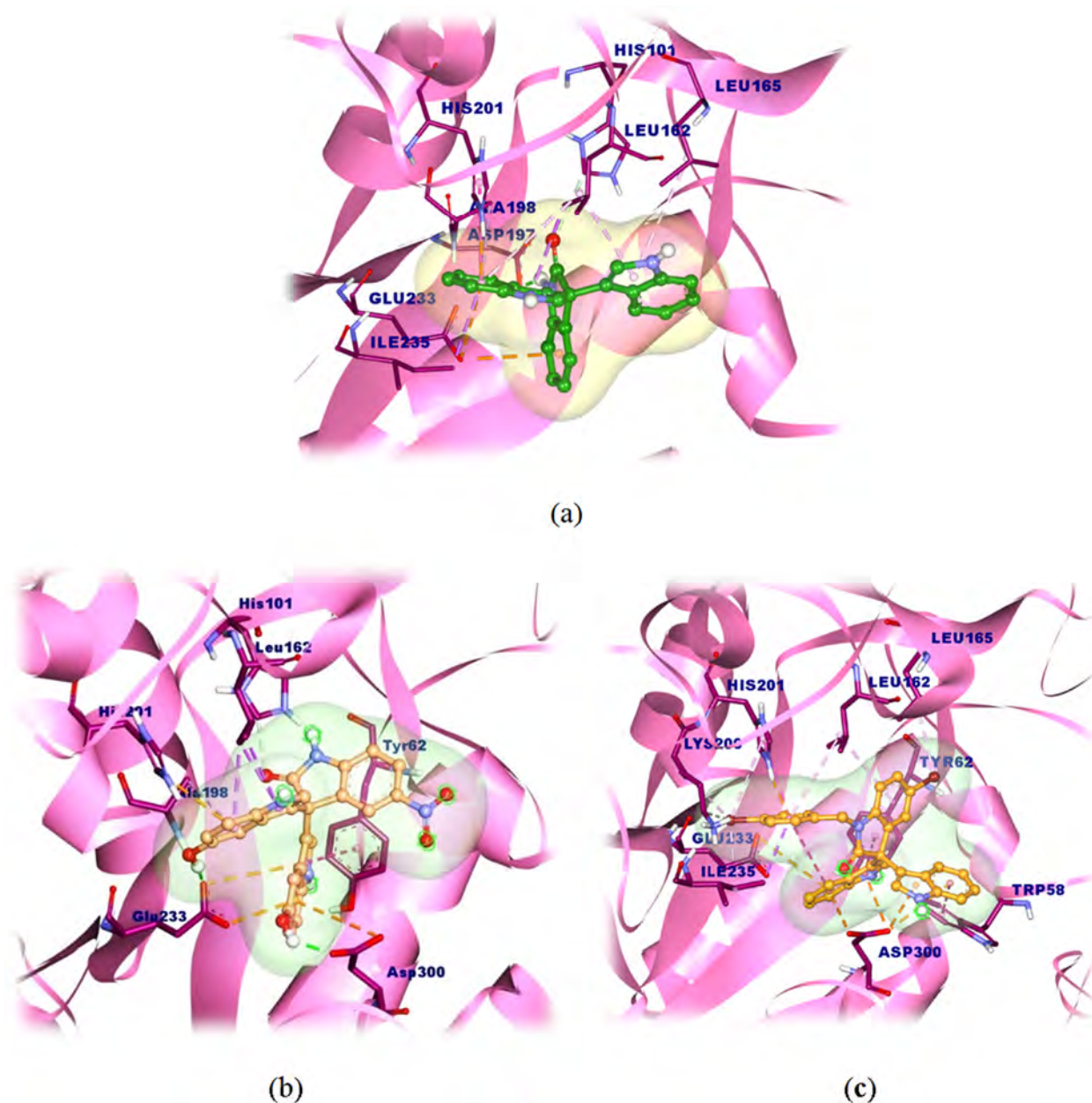


Figure 2. Binding interaction of (a) indolin-2-one **1a**, (b) indolin-2-one **1i**, and (c) indolin-2-one **1n** with α -amylase (PDB ID: 6OCN).

1j increased its activity to almost five-fold (79 ± 5 vs 16 ± 6 , Table 1, entry 10 vs entry 1). At this stage, we predicted that the introduction of *N*-alkyl substituents to **1j** will increase its inhibition activities. However, while compound **1k** with methyl substituents showed a moderate increase in the % inhibition activity to 86 ± 6 , compound **1l** with benzyl substituent decreased the % inhibition activity to 76 ± 8 perhaps due to steric effects which prevented better fitting with the enzyme active sites. A significant increase in the activity occurred by changing the strong electron-withdrawing NO_2 group of **1l** to bromine as in **1m** which showed % inhibition of 92 ± 3 , or related **1n** which showed % inhibition of 94 ± 3 . Replacement of bromine of **1f** with chlorine to get **1o** resulted in a small increase in the % inhibition activity (Table 1, entry 6 vs entry 15). Methylation of **1o** with to obtain **1p** did not increase the % inhibition activity.

3.2.2. α -Amylase inhibition activities

The tested compounds in Table 1 showed high to very high α -amylase % inhibition activities ranging from 72 ± 5 to 92 ± 4 , except **1i** which

showed a % inhibition value of 51 ± 4 , all measured at a concentration of $50 \mu\text{g/ml}$. Overall, the differences in activities between structurally related compounds are not as pronounced as in the α -glucosidase case. Interestingly, compound **1a** which showed the lowest α -glucosidase inhibitory activity exhibited one of the highest α -amylase % inhibition activity of 92 ± 4 . The inhibition activity varied with the size/number of the substituents. For example, as the size/number of the substituents increases in compounds **1a-e**, the corresponding inhibition values decrease (Table 1, entries 1–5). The same is also observed for compounds **1f** and **1g**. The situation becomes ambiguous when we consider the effects of electron-withdrawing and donating substituents which gave no obvious trend. For example, the parent compound **1a** showed very high % inhibition activity similar to compounds **1j** with 93 ± 6 and **1k** with 91 ± 5 having strong electron-withdrawing group NO_2 and donating methyl moieties. The inhibition values in Table 1 suggest that compounds **1a-h** and **1j-p** have a similar mode of interactions with α -amylase enzymes and that mode is different in the case of **1i**. As mentioned in the introduction section, lower α -amylase inhibition is desired to overcome the

gastrointestinal side-effects. Therefore, compound **1i** serves this purpose considering that it has the lowest α -amylase activity and high α -glucosidase activity.

3.3. Molecular docking studies

Docking simulations were performed on 3,3-di(indolyl)indolin-2-ones **1a**, **1i**, and **1n** to predict the binding interaction of these compounds in the active site of both enzymes. These compounds were selected because they showed the highest contrasting α -glucosidase and α -amylase inhibition values (Table 1, entries 1, 9, and 14). In the docking simulation, all the 3,3-di(indolyl)indolin-2-ones **1a**, **1i**, and **1n** recognized the binding pocket of both enzymes correctly. These indolin-2-ones formed stable key interactions with the active sites of both enzymes.

In the case of α -glucosidase, the binding affinities of the three indolin-2-ones is in the order **1a** (−7.45 kcal/mol) > **1i** (−7.84 kcal/mol) > **1n** (−8.26 kcal/mol) which is consistent with the experimental % inhibition trend values $16 \pm 6 < 67 \pm 13 < 94 \pm 3$, respectively. The theoretical binding modes of compounds **1a**, **1i**, and **1n** with α -glucosidase are shown in Figure 1. Indolin-2-one **1a** formed hydrophobic π - σ interaction with the Trp376. One of the indole rings formed π - π T-shaped interaction with Phe649 and Trp481 while the other indole rings formed π -alkyl with Leu650 and Ala284. Both of the indole rings also formed π -anion interaction with Asp616. From the docking analysis, hydrogen bonds were observed between NH of each indole rings with Met519 (bond length: 2.93 Å) and Asp282 (bond length: 1.88 Å). Indolin-2-one **1i** which has an NO₂ and OH groups formed new hydrogen bonds with Leu677 (bond length: 2.34 Å) and Asp404 (bond length: 2.76 Å), respectively, while maintaining the hydrogen bond between the NH of the other indole rings and Asp282 (bond length: 2.06 Å). Additionally, the NO₂ group not only formed a conventional hydrogen bond, but also a non-classical or carbon-hydrogen bond with the Ser676. The OH group formed π -lone pair interaction with Trp376. However, it was different from indolin-2-one **1n** which did not show the presence of any hydrogen bonds. The *p*-bromobenzyl group on the indolin-2-one ring in indolin-2-one **1n** formed hydrophobic π - π stacking and π -alkyl interactions with Phe649, His674, Trp613, and Trp516 residues. In addition, the benzene ring of the *p*-bromobenzyl group formed a π -anion interaction with Asp518. Besides, both indolin-2-ones **1i** and **1n** maintained the hydrophobic interaction with Trp376, Leu650, and Ala284 residues.

In the case of docking with α -amylase, the binding affinities of indolin-2-ones were in the order **1i** (−8.42 kcal/mol) > **1a** (−8.47 kcal/mol) > **1n** (−8.75 kcal/mol) which is different from the trend with α -glucosidase. The binding modes of indolin-2-ones **1a**, **1i**, and **1n** with α -amylase are shown in Figure 2. The protein-ligand complex analysis of compound **1a** showed there were two kinds of hydrogen bonds observed on the indolin-2-one ring which were conventional and non-classical hydrogen bonds. The NH of the indolin-2-one ring formed the conventional hydrogen bond with Asp197 (bond length: 1.74 Å) while the carbonyl group of the indolin-2-one ring formed the non-classical hydrogen bond with His101 (bond length: 3.52 Å). The indole rings of compound **1a** formed several hydrophobic interactions which were π - σ interaction with Ile235 and Leu162, π -alkyl interaction with Ala198, Leu165, and Leu162. Moreover, there were electrostatic interactions via a π -cation interaction between the indolin-2-one ring and Glu233 then a π -anion interaction between one of the indole rings and His201. From the docking analysis, the carbonyl group of the indolin-2-one ring in compound **1i** formed a non-classical hydrogen bond with His101 (bond length: 3.02 Å). One of the indole rings of compound **1i** formed π -alkyl, π - σ , and π -anion interactions with Ala198, Leu162, and His 201, respectively. The other indole rings formed π - π T-shaped and π -anion interactions with Tyr62 and Glu233, respectively. Meanwhile, the substitution of the OH group in the indole rings formed hydrogen bonds with Glu233 (bond length: 1.72 Å) and Asp300 (bond length: 1.83 Å). In this receptor, again, compound **1n** showed no hydrogen bond but had five types of hydrophobic interaction in its protein-ligand complex. The

bromo group of the indolin-2-one ring formed an alkyl hydrophobic interaction with Leu165, while bromo group of the *p*-bromobenzyl of the indolin-2-one ring formed alkyl and π -alkyl interaction with Ile235 and His201, respectively. The *p*-bromobenzyl group also formed π - σ and π -cation interaction with Ile235 and His201, respectively. The indole rings of compound **1n** formed π - π stacking and π - π T-shaped interactions with Trp58 and Tyr62, respectively. Also, the indole rings formed π -anion interaction with Asp300 and Glu233.

4. Conclusions

We have synthesized a series of 3,3-di(indolyl)indolin-2-ones **1a-p** and examined their α -glucosidase and α -amylase inhibitory activities. Overall, the compounds showed desired higher α -glucosidase activities and desired lower α -amylase activities than standard drug AGI acarbose. The inhibitory activity against α -glucosidase varied greatly in comparison to α -amylase concerning the substituents on the core structure. Molecular docking studies showed that the tested compounds interacted with the active sites of both α -glucosidase and α -amylase and the trend in the binding energy values parallel with the % inhibition values. The results suggest that 3,3-di(indolyl)indolin-2-ones, especially indolin-2-one **1i**, are promising AGIs.

Declarations

Author contribution statement

Mardi Santoso, Azminah Azminah, Zaher M. A. Judeh: Conceived and design the experiments; Contributed reagents, materials, analysis tools or data; Wrote the paper.

Li Lin Ong, Nur Pasca Aijijiyah, First Ambar Wati, Rose Malina Annuur, Arif Fadlan: Performed the experiments; Analyzed and interpreted the data.

Funding statement

This work was supported by Ministry of Education, Culture, Research, and Technology, Indonesia (WCP Program, PDUPT grant no. 925/PKS/ITS/2021) and Nanyang Technological University (RG142/16).

Data availability statement

Data will be made available on request.

Declaration of interests statement

The authors declare no conflict of interest.

Additional information

No additional information is available for this paper.


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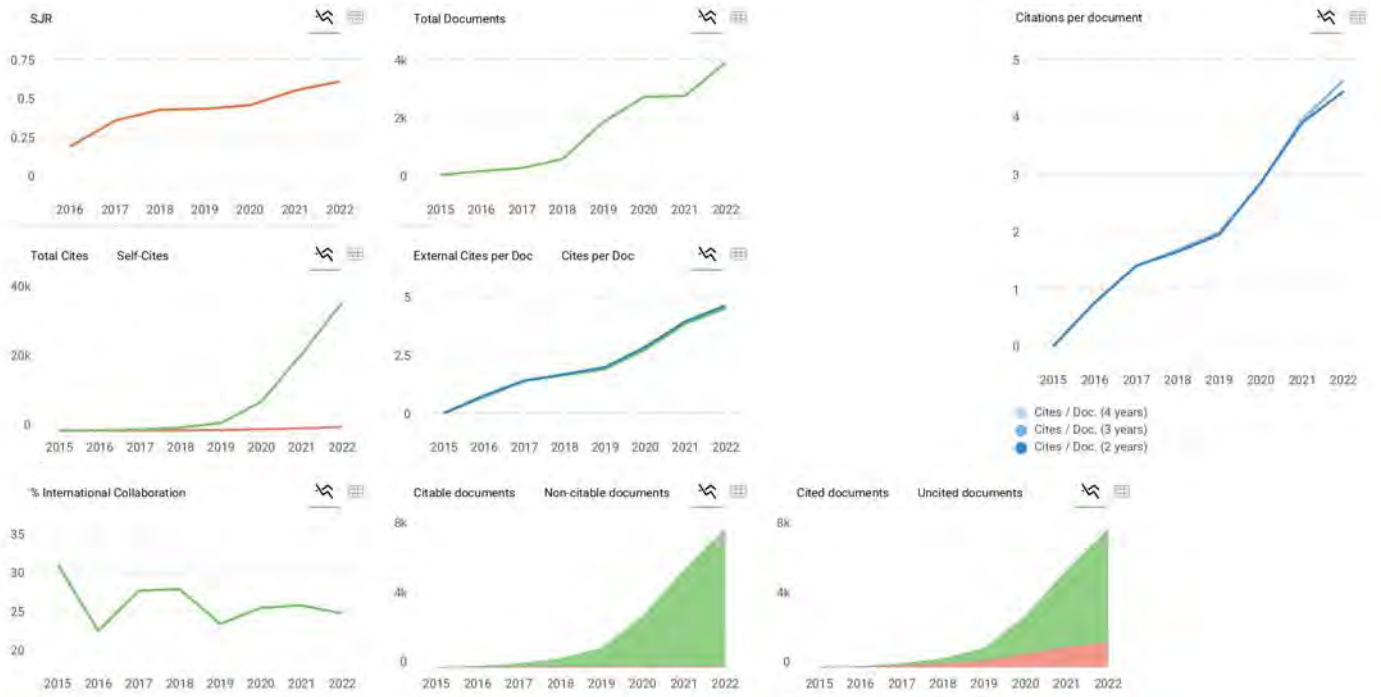
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